



Diffusion and Bacteria:

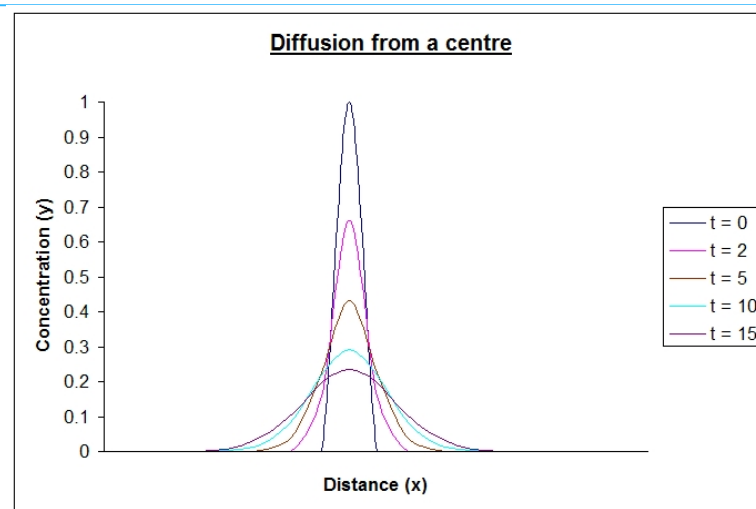
A Study in Biophysics

Walter Brose



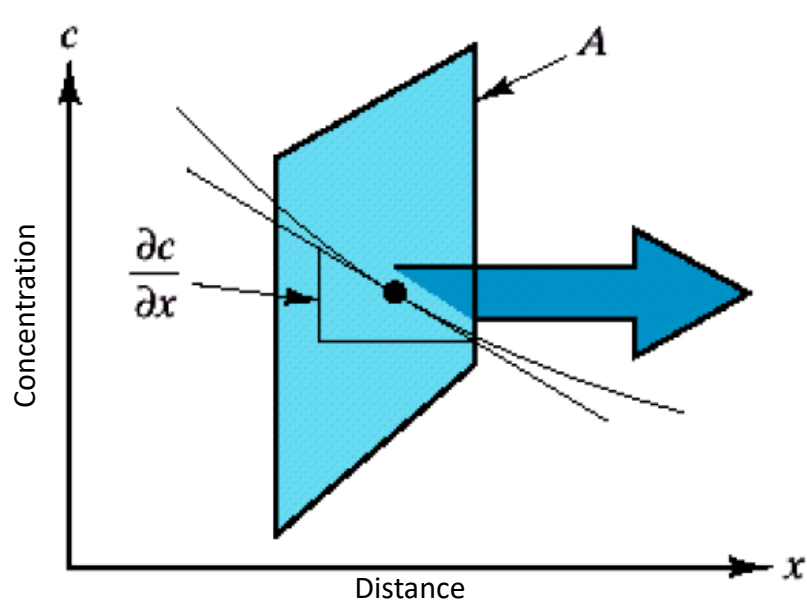
Diffusion

- Statistical probability of particles moving from high concentration to low concentration



Fick's Law

- Relationship between Flux (j) and Slope of Concentration Gradient (dc/dx)



$$j = -D \frac{dc}{dx}$$

Diffusion Coefficients

- Physical property of a binary system
- Rate of area per time (dimensionally)
- Similar to Friction Coefficients
- Ranges for different binary systems
 - Gas – Gas = $10^{-5} \text{ m}^2/\text{s}$
 - Gas – Liquid = $10^{-9} \text{ m}^2/\text{s}$
 - Gas – Solid = $10^{-13} \text{ m}^2/\text{s}$
- Dimensionally
 - $D = Lv$

Bacteria

- Utilize *Passive Transport* (Diffusion) to satisfy life functions
 - This places an upper limit on the size bacteria can be
 - Too big = Diffusion too slow to keep alive

Maximum Size of Bacteria

- Equate *Intake Rate* (I) to *Metabolic Rate* (M)
 - I – the rate at which O_2 can be taken into the bacterium through outer shell
 - M – the rate at which the bacterium can metabolize the O_2 through whole volume

$$I = \text{Flux } (j(r)) \times \text{Surface Area of Sphere } (4\pi r^2)$$

$$M = \text{Experimental Consumption Rate } (\alpha) \times \text{Volume of Sphere } (\frac{4}{3}\pi r^3)$$

- Solve for Radius

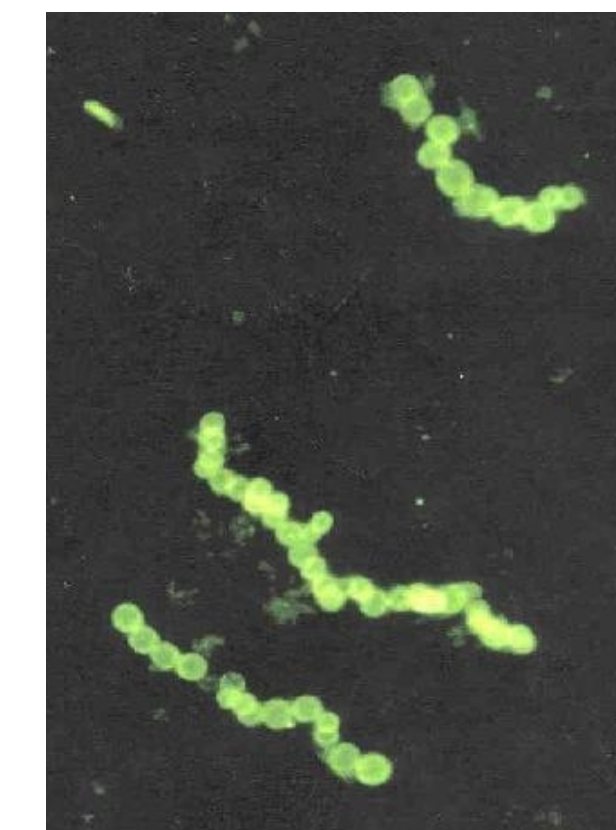
$$R = \sqrt{\frac{C_0(3)D}{\alpha}}$$

- Here there is only have dependence on constants or experimentally determined values

- Plugging in values I get

$$R = \sqrt{\frac{(.706 \frac{\text{mol}}{\text{m}^3}) 3 (2.5 \times 10^{-9} \frac{\text{m}^2}{\text{s}})}{1.1 \frac{\text{mol}}{\text{m}^3 \text{s}}} = 69.3 \mu\text{m}}$$

Thiomargarita Namibiensis



- Largest bacteria ever found
- Radius range: $50\text{-}150 \mu\text{m}$
- My calculation fits into this range
- Experimental values not "idealized" so my number on smaller end

Conclusion

- Diffusion is random molecular motion that follows a statistical probability
- Fick's Law, using concentrations and diffusion coefficients, in combination with known consumption rates, allows us to calculate upper size limit of bacteria
- This requires no biochemistry, only the physics of diffusion!

Works Cited

- Levin, P. A., & Angert, E. R. (2015). Small but mighty: Cell size and bacteria. *Cold Spring Harbor Perspectives in Biology*, 7(7), a019216. doi:10.1101/cshperspect.a019216
- Marshall, W. F., Young, K. D., Swaffer, M., Wood, E., Nurse, P., Kimura, A., . . . Roeder, A. H. K. (2012). What determines cell size? *BMC Biology*, 10(1), 101. doi:10.1186/1741-7007-10-101
- Maximum dissolved oxygen concentration saturation table. Retrieved from <https://dnr.mo.gov/env/esp/wqm/DOSaturationTable.htm>
- Nelson, P. (2014). *Biological physics* (5. print. ed.). New York, NY: Freeman.
- Nuclear Waste.20.3: *Applications of diffusion*
- Philip Nelson. (2003). *Biological physics: Energy, information, life* (5th ed.)
- Russel, W. B. (1981). Brownian motion of small particles suspended in liquids. *Annual Review of Fluid Mechanics*, 13(1), 425-455. doi:10.1146/annurev.fl.13.010181.002233